The Free Radical Scavenging and Tyrosinase Inhibition Activities of Protocatehuic Acid and Caffeic Acid Derivatives

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Objects

The most current research on antioxidant action focuses on phenolic compounds. They are the most common compounds in fruits and vegetables and have a strong antioxidant capacity. Caffeic acid (3, 4-dihydroxycinnamic acid) is among the major hydroxycinnamic acids present in many kinds of fruits. Protocatehuic acid belongs to the hydroxybenzoic acid has also been identified as one of the active antioxidants. In the present study, the antioxidant properties of the caffeic acid and protocatehuic acid derivatives were synthesized and evaluated by using different in vitro antioxidant assays such as 1.1-diphenyl-2-picryl-hydrazyl free radical (DPPH•) scavenging and DNA protection activities. The effective antioxidants were also surveyed of the tyrosinase inhibition activity.

Keywords free radical scavenging, tyrosinase inhibition, protocatehuic acid, caffeic acid, esterification, amidization

Materials and methods

Esterification and amidization reactions for the syntheses of the protocatehuic acid and caffeic acid derivatives were illustrated in Figure 1.

(A) Esterification

HO. HO

 $\underline{1}$ R = Me; $\underline{2}$ R = Et; $\underline{3}$ R = Pr; $\underline{4}$ R = *i*-Pr; $\underline{5}$ R = CH₂Ph; $\underline{6}$ R = (CH₂)₂Ph; $\underline{7}$ R = (CH₂)₂Ph $\underline{\mathbf{8}} \operatorname{R} = \operatorname{CH}_2\operatorname{Ph}; \underline{\mathbf{9}} \operatorname{R} = (\operatorname{CH}_2)_2\operatorname{Ph}; \underline{\mathbf{10}} \operatorname{R} = (\operatorname{CH}_2)_2\operatorname{Ph}$

(B) Amidization

$$HO \longrightarrow f \longrightarrow n + RNH_2 \xrightarrow{DCC} HO \longrightarrow f \longrightarrow n + RNH_2 \xrightarrow{DCC} HO \longrightarrow f \longrightarrow n + RNH_2 \xrightarrow{DCC} HO \longrightarrow f \longrightarrow n + RNH_2 \xrightarrow{O} R + RNH$$

Caffeic acid n = 1

n = 0 Protocatechuic acid:

n = 1 Caffeic acid

n=1 14 R = Hex; 15 R = Hept; 16 R = Oct

acid and caffeic acid derivatives. (A) Esterification reactions for compounds 1-7 (derived from protocatehuic acid) and 8-10 (derived from caffeic acid) (B)Amidization for compounds 11-13 (derived from protocatehuic acid) and 14-16 (derived from caffeic acid). DCC : N. N'- Dicvclohexyl carbodiimide : THF : Tetrahydrofuran

Figure 1. Illustration of the chemical reactions for the syntheses of the protocatehuic



Conclusion

As seen in results, compounds 1, 2, 5, 6 (esterification derivatives of protocatehuic acid) had effective DPPH• scavenging activity in a concentration dependent manner (1–50 μ g/mL). There is a significant decrease in the concentration of DPPH• due to the scavenging capacity of these compounds. Present study has clearly shown that two protocatehuic acid derivatives, compounds 1 (methyl 3, 4- dihydroxy benzoate) and 2 (ethyl 3, 4- dihydroxy benzoate), were effective antioxidants in different in vitro antioxidant assays including total antioxidant activity by DPPH. scavenging and DNA protection assays. In addition, compounds 6 and 7 derived from protocatehuic acid were potent tyrosinase inhibitors.



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Figure 2. Tyrosinase inhibition activities of the antioxidants. Esterification and amidization derivatives of the protocatebuic acid (compounds 1-7, 11-13) and caffeic acid (compounds 8-10. 14-16) were assayed for the muslicoom tyrosinase inhibition activities. Native protocatehnic acid (Pr), caffeic acid (Ca), vitamin C (Vc), kojic acid (Ko) were used as control. The ICso values of the inhibition are shown at the concentrations of pg ml. Note the compounds 6 and 7 belong to protocatehnic acid derivatives showed potent tyrosinase inhibition activities

Compound 6

Free Radical Scavenging Activity