准考證號碼:

嘉南藥理科技大學九十四學年度碩士班考試入學招生

有機化學試題(藥物科技研究所碩士班甲組、乙組) 本試題共1張2面

一、選擇題(45%)選擇答案欄

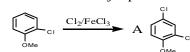
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	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15

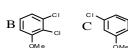
1.A compound of formula C₉H₁₀ has the ¹H NMR spectrum indicated below. What is the structure of the compound? δ 7.13, 4H singlet, δ 2.89, 4H triplet, δ 2.04, 2H multiplet

2. Which is this polysaccharide below? A. amylose B. amylpectin C. cellulose d. glycogen

- 3. Which compound will absorb ultraviolet radiation at the longest wavelength?

4. Predict the major product of the following reaction:

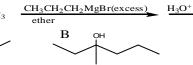


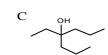




E.No reaction occurs

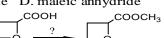
- 5. What is the major organic product from this series of reactions:
 - CH₃CH₂CH₂COOCH₃



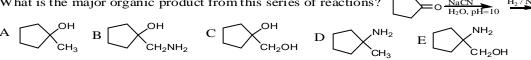


6. Phenobarbital $(C_{12}H_{12}N_2O_3)$, a widely prescribed sedative in the 1950's and 1960's, can be prepared by the reaction of urea with diethyl 2-ethyl-2-phenyl propanoate. Which is phenobarbital?

- 7. What is the name of this compound?
- A. cyclopentane anhydride B. succinic anhydride C.malonic anhydride D. maleic anhydride
- 8. Which reagent should give the highest percent yield for this reaction? A. CH_3OH , H^+ B. $SOCl_2$, then CH_3OH C. CH_2N_2 , ether D. $NaOCH_3$



9. What is the major organic product from this series of reactions?



- 10. What is the major organic product from the following reactions? 18OH
- ¹⁸OH 11. Which of the following would be appropriate solvents for preparing an organolithium reagent? I. $CH_3CH_2CH_2CH_2CH_2OH$ II. $CH_3CH_2OCH_2CH_2CH_3$ III. $CH_3CH(OH)CH_2CH_2OH$ IV. $CH_3OCH_2CH_2OCH_3$ A. I, II B. I, III C. I, IV D. II, III E. II, IV F. III, IV

12. Which has the highest boiling point?

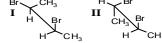
A. $CH_3CH_2CH_2CH_3$ B. $CH_3CH_2CH_2OH$ C. $HOCH_2CH_2OH$ D. $CH_3CH_2OCH_3$ E. $CH_3CH_2CH_2F$

- 13. What is the name of this compound? H₃C
 - B. 3(Z)-penten-2(R)-ol C. 3(E)-penten-2(S)-ol D.3(Z)-penten-2(S)-ol A. 3(E)-penten-2(R)-ol

14. What is the relationship between **I** and **II**?

A. diastereomers B. enantiomers C. constitutional isomers

D. different conformations of same molecule E. diastereomers



15. Starting with the weakest base, arrange the following in order of increasing basicity.

B. II, I, IV, V, III C. IV, I, II, III, V D. IV, I, III, II, V

1. Rank the following comp	pounds in each set in order of	of increasing acid strength	: (9 %)
(1) benzoic acid, p-meth	nylbenzoic acid, m-methylbe	enzoic acid, p-nitrobenzoio	e acid, m-nitrobenzoic acid
(2) phenol, p-methylphe	nol, m-methylphenol, p-chl	orophenol, m-chloropheno	ol
(3) ethanoic acid, chloro	ethanoic acid, dichloroetha	noic acid, trichloroethanoi	c acid
2. Draw a structural formul	las that corresponds to each	of the following names: (8 %)
(1) Pyrimidine	(2) Pyridine	(3) Pyrrole	(4)Anisole
2 Outline all stone in the ex	enthacia of the following cor	nnounds from honzono, us	ing any reagants (20.9%)
(1) 3, 5-Dibromo-2-met	nthesis of the following cor	_	chloroacetophenone
(3) 2- Amino-4- bromo-	-5-chlorobenzoic acid	(4) p-Chloros	tyrene
4. Explain the meaning for (1) Bimolecular nucleopl	each of the following terms: hilic substitution (SN ₁)	(6 %) (2) Crossed Clais	en condensation
(3) Aldol Condensation			
5. Arrange the following all (1) pentane, hexane, octar	kanes in each set in order of ne, decane, heptane (4 %) utane, 2, 2-dimethylbutane, 2, 3-dimethylbutane
6. Give typical infrared (IR	a) absorption bands for each	function group: (8 %)	
(1) C-H	$(2) C \equiv N$	(3) C - O	(4) Ar - H

二、問答題:(55%)