

# 臺南藥理學院專題研究計畫成果報告

## 酒精性飲料中胺基甲酸乙酯之簡易快速氣相層析法

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### 摘要

市售酒精本研究發展了飲料中胺基甲酸乙酯(ethyl carbamate, EC)之快速氣相層析(使用 CP-wax, 30m X 0.53mm)直接注入分析法。酒精飲料樣品經簡易之前處理包括:(一)直接取樣(不做任何處理)注入法(Method I)、(二)樣品經濃縮後(濃縮 5-10 倍)之注入法(Method II)、及(三)樣品中之 EC 經乙酸乙酯(v/v=1/2)萃出濃縮後之注入法 (Method III)。選用 2- 甲基- 戊酸(2-methyl-n-pentanoic acid)為內標準品，分別添加  $5 \mu\text{g}$  之胺基甲酸乙酯於多種蒸餾酒及非蒸餾酒中，Method I 、Method II 及 Method III 之回收率分別為 95-119% 、96-112% 及 96-106%，變異係數 CV% 均在 9.5% 以下，顯示本研究方法相當精確。以本方法分析市售蒸餾酒 21 件及非蒸餾酒 24 件之胺基甲酸乙酯含量。Method I 、Method II 及 Method III 對上述 45 件樣品中胺基甲酸乙酯之檢出率分別達 22 、72 及 77%。其中以 Method III 分析之結果顯示 45 件酒精性飲料中胺基甲酸乙酯之含量分別為  $0 \sim 2.74$  、 $0 \sim 2.28$  及  $0 \sim 2.57 \mu\text{g/mL}$ 。

關鍵字：酒精性飲料、胺基甲酸乙酯、直接注入法、氣相層析法、定量分析。

## 前言

胺基甲酸乙酯 (Ethyl carbamate,EC) , 又稱 Urethane , 自然存在於酒精性飲料及一般醱酵食品中。胺基甲酸乙酯具生物活性，在動物體中會導致腫瘤產生。在 1970 年初，陸續有不少的研究報告指出在各種發酵食品，尤其是酒精性飲料中含有 EC。由於 EC 可能導致癌症，因此在近幾年來有關酒類中 EC 含量之問題頗受重視。在 1985 年之前，各國對於酒類中 EC 含量並無任何規定，但由於加拿大的酒類曾被發現含高量 EC，因此由其健康保護局訂定各種酒類之 EC 含量標準，餐酒 (table wine) 中 EC 含量不可超過 30 ppb；葡萄酒中 EC 含量不可超過 100 ppb；蒸餾酒中 EC 含量不可超過 150 ppb，而白蘭地等 EC 含量不可超過 400 ppb。同時在 1985 年下令由市場回收含 EC 超過規定量 (150ppb) 之某些廠牌之酒類。另外也在其他食品如牛乳、麵包、柑橘果汁....中發現 EC 含量。Zimmerli 等學者研究估計，成人每日 EC 之攝食量約為 20ug/Kg body weight。

由於 EC 具有潛在危害性，近年來頗受歐美各國之重視，紛紛致力於研究其在食品中之含量、分析方法、形成原因及降低含量之方法，以減少其對消費者健康之毒害性。

國內傳統之各式酒精性飲料種類相當多，加上近年來酒類之開放進口，國人對於酒精性飲料之消費量非常高，但是對其中可能存在的 EC 含量，卻未能引起政府相關單位及消費大眾之廣泛注意。因此為了維護國人飲食之安全，實有必要建立國內此類食品中 (及其他含 EC 之發酵食品) EC 含量之基本衛生資料。然而目前 EC 之分析方法多以溶劑 (二氯甲烷或乙醚) 萃取法或以二相透析萃取法、萃取後以 GC 分析定量。但由於 EC 為偏親水性物質，萃取率及回收率均偏低，且萃取時的濃度稀釋、濃縮過程造成的分解及管壁殘留、衍生反應的反應率等等，都可能造成回收率精確度偏低，而且這些前處理過程相當耗時耗工，不適合作為例行分析工作之方法。

本人曾以 splitless direct injection 之氣相層析方法，成功的建立了食品中咖啡因、膽固醇及各種防腐劑之簡易快速定量方法。基於此本研究擬採用 splitless direct injection 之方式，探討以 GC 直接分析 (或濃縮後再分析) 各種酒精性飲料中 EC 含量之可行性。以期建立發酵食品中 EC 之簡易快速定量分析方法，可供作為 EC 之例行分析之標準方法。

## 材料與方法

## 一、材料：

市售各種台灣製及進口之酒精性飲料包括蒸餾酒(distilled spirit) 21件及非蒸餾酒(non-distilled spirit)24件共54件，購自台南與屏東地區之超級市場；胺基甲酸乙酯(ethyl carbamate, EC)、癒創木酚(guaiacol)、1,3-二醇(1,3-butanediol)、1,5-戊二醇(1,5-pentanediol)、3-甲基-1,5-戊二醇(3-methyl-1,5-pentanediol)、2-甲基-正戊酸(2-methyl—pentanoic acid)及1,6-己二醇(1,6-hexanediol)等標準品(純度均在98以上)為日本東京化成SCI公司商品製造。

## 二、方法：

### (一) 標準品及內標準溶液之配製：

分別精稱 0.4g 之胺基甲酸乙酯及 2- 甲基 - 正戊酸 (2-methyl-n-pentanoic acid) 於 100mL 定量瓶中，以甲醇稀釋定量至標線。

(二) 胍基甲酸乙酯對2-甲基-正戊酸之相對表現因子(relative response factor, RRF)之測定：

將0.4% (w/v)之胺基甲酸乙酯與0.4% (w/v)之2-甲基-正戊酸(內標準品) 甲醇溶液依不同比例混合(ethyl carbamate /2-methyl-pentanoic acid=2/1、1/1及1/2)，依氣相層析分析，

以兩者在氣相層析儀之波峰面積，計算胺基甲酸乙酯對2-甲基-正戊酸之相對表現因子。亦即：

$A_s$  = ethyl carbamate 之 GC 波峰面積；  $A_{IS}$  = 2-methyl-pentanoic acid 之 GC 波峰面積；  $W_s$  = ethyl carbamate 之重量(ug)；  $W_{IS}$  = 2-methyl-pentanoic acid 之重量(ug)。

### (三) 胺基甲酸乙酯之定量分析：

\*酒精飲料樣品經簡易之前處理包括：

- (1)直接取樣(不做任何處理)注入法(Method I) 、

(2)樣品經濃縮後(濃縮 5-10 倍)之注入法(Method II) 、及

(3)樣品中之 EC 經乙酸乙酯( $v/v=1/2$ )萃出濃縮後之注入法 (Method III) 。

即直接取 $0.1 \mu\text{L}$ 注入GC中分析。胺基甲酸乙酯之含量可依公式(2)  
計算求得：

$$\text{Ethyl carbamate (ug/mL)} = (\text{A}_S/\text{A}_{IS}) \times (\text{W}_{IS}/\text{RRF}) \times 1/V \quad \dots\dots(2)$$

$V$  = 檢體之體積(mL)。

### Method I :直接注入法



## **Method II :濃縮後注入法**

取酒類樣品 25 mL



放入 250 mL 濃縮瓶中



加入 400  $\mu\text{g}/100 \mu\text{L}$  之內標準溶液

(IS = 2-Methyl-n-Pentanoic Acid)



減壓濃縮至約 2-5 mL



取 0.1  $\mu\text{L}$  注入氣相層析儀

分析 EC 含量

## **Method III :溶劑萃取法**

取酒類樣品 25mL



放入 100 mL 量筒中



加入 200  $\mu\text{g}/50 \mu\text{L}$  之內標準溶液

(IS = 2-Methyl-n-Pentanoic Acid)



以 50 mL 之乙酸乙酯萃取 2-3 次



乙酸乙酯萃取液減壓濃縮至約 0.2-0.5 mL



取 0.1  $\mu\text{L}$  注入氣相層析儀

分析 EC 含量

#### (四)最低檢出量試驗：

以胺基甲酸乙酯標準溶液( $200 \mu\text{g/mL}$ )，分別以純水稀釋至 2.5、1.0、0.5、0.25及 $0.1 \mu\text{g/mL}$ ，各取1 mL上述溶液，添加 $50 \mu\text{g}$ 之內標準溶液(2-methyl-pentanoic acid)，混合均勻後，直接以GC分析之，估計胺基甲酸乙酯之最低檢出量，每一濃度均做三重覆。

#### (五)添加回收試驗：

分別添加 $5 \mu\text{g}$ 之胺基甲酸乙酯於1 mL, 25 mL及25mL之蒸餾酒及非蒸餾酒中，再分別添加200, 400及 $200 \mu\text{g}$ 之內標準溶液，經混合均勻後，分別以Method I, Method II 及Method III分析之，每一添加量均做三重覆，同時作空白試驗，計算胺基甲酸乙酯之回收率。

#### (六)氣相層析儀條件：

GC儀器：Hitachi G-3000； FID 檢測器： $\text{H}_2$ ，flow rate= $30 \text{ mL/min}$ ，air flow rate= $300 \text{ mL/min}$ ； 檢測器溫度= $280^\circ\text{C}$ ；注入口溫度= $240^\circ\text{C}$ ； 攜帶氣體： $\text{He}$ ，flow rate =  $4 \text{ mL/min}$ ； 分離管柱：CP-Wax ( $30 \text{ m} \times 0.53 \text{ mm}$ ,  $1 \mu\text{m}$ )，(Chrompack, Netherlands)；烘箱溫度：起始溫 $100^\circ\text{C}$ ，2鐘後，以每分鐘 $3^\circ\text{C}$ 之速率升溫至 $135^\circ\text{C}$ ，再以每分鐘 $5^\circ\text{C}$ 之速率升溫至 $155^\circ\text{C}$ ，再以每分鐘 $50^\circ\text{C}$ 之速率升溫至 $300^\circ\text{C}$ ； 注入量：  $0.1 \mu\text{L}$ ； 注入型式： direct injection mode。

## 結果與討論

### 一、氣相層析條件之探討：

最近有關胺基甲酸乙酯之傳統前處理之分析定量方法，均是以有毒之有機溶劑如(二氯甲烷乙醚)多次的連續萃取，再經濃縮後以GC分析定量之，此傳統之前處理步驟。其中李等<sup>(19)</sup>是以1 mL醬油，調pH至2.0後，以

10倍體積之乙醚連續萃取三次，乙醚層經濃縮至1 mL後，再以GC分析定量之（採split mode 注入方式）。此法每分析一個樣品耗時約30~40min，果糖酸之回收率約在90%左右。而本研究室先前<sup>(20)</sup>以5 mL醬油，調pH至2.0再以NaCl飽和後，以6倍以上乙酸乙酯萃取，加入內標準品（heptanoic acid）後，再稍經減壓濃縮去除部分乙酸乙酯，即以GC分析定量之（採splitless direct injection mode 注入方式）。此法每分析一個樣品需時約20min，果糖酸之回收率在92~98%左右。以上兩種分析果糖酸之方法，已是相當簡便快速，缺點是果糖酸仍是親水性極高化合物，需以大量之有機溶劑萃取才可達到90%以上之回收率。Sheu et al., 1996, 發展出二相透析萃取法，此法相當費時，且回收率亦不高。

氣相層析具有高解析度及高靈敏度等優點，仍為現代最重要分析技術之一。本研究室從事多年GC分析發現，可於注入口(injector) 之insert 玻璃管內塞上玻璃棉或接上一小段（約1~2m）保護管柱（guard column），可有效防止非揮發性之有機物等進入分離管柱，以降低污染物之干擾，減少波峰之拖尾現象（tailing），提高GC層析圖之解析度<sup>(22-24)</sup>。本研究室亦發現市面販售之GC分離管柱（megapore column）之耐水性相當高<sup>(24)</sup>。即使直接注入水溶液之樣品於GC中，GC管柱之分離效果與滯留時間之再現性仍與原先之新管柱一樣。且滯留在insert玻璃管內之鹽分及雜質不需經常的清洗，約可在分析100個以上之樣品後才需清除，而insert之清除亦很簡便，只需將insert玻璃管取出，用鹽酸浸泡10分鐘後，將玻璃棉取出，用水沖洗乾淨，乾燥後換上新的玻璃棉即可放回injector內使用。

基於上述之理由，本研究擬將酒精性飲料，(一)不經任何前處理，加入適當適量之內標準水溶液後，直接注入氣相層析儀中(Method I)，(二)樣品經濃縮後注入 GC(Method II)(三)樣品以乙酸乙酯萃出濃縮後注入 GC(Method III)配合適當的分離管柱及氣相層析條件，以期建立簡便、快速又精確之酒精性飲料中胺基甲酸乙酯之定量分析方法。

本研究之結果顯示，在分離管柱之選擇方面，在分析極性之胺基甲酸乙酯(EC)乃採用極性之管柱如CP-Wax為宜。樣品採用直接注入方式進行分析，管柱分析之昇溫程序為起始溫度為100°C經2 min後，每分鐘昇溫8°C至220°C，再以每分鐘50 °C之速率升溫至250 °C。檢測器之溫度為290°C，注入口溫度為250°C。依上述條件直接注入酒精性飲料而得EC之GC滯留時間為12.73min，如圖一至圖四所示，分析一個樣品需約20 min。

在內標準品的選擇方面，於酒精性飲料等樣品中逐一加入少量之 guaiacol ， 1,3-butanidiol, 2-methyl-n-pentanoic acid, 1,5-pentanediol 、

3-methyl-1,5-pentanediol及1,6-hexanediol 六種水溶性之標準品，然後依上述 GC 條件分析之，由分析之結果顯示，1,5-pentanediol、3-methyl-1,5-pentanediol 及 1,6-hexanediol 之滯留時間分別為 11.41, 14.32, 14.68, 17.24, 18.57 及 19.08 min，如表一所示。經與空白樣品之 GC 圖比對，以 2-methyl-n-pentanoic acid 最適宜，波峰完全不會與樣品中之波峰重疊，因此選定為本研究中 EC 分析定量之內標準品。

依上述之 GC 分析條件，Method I, Method II 及 Method III 三種 分析市售酒類樣品，分析一個樣品僅需 20 分鐘左右，如圖一～圖五所示。

### 三、 胺基甲酸乙酯對內標準品之相對表現因子：

本研究選擇水溶性之 2-methyl-n-pentanoic acid 作為直接定量酒類等水溶液樣品中 EC 之內標準，要有準確之定量，必需先定出 EC 對 2-methyl-n-pentanoic acid (內標準) 之 相對表現因子(RRF)，則樣品中 EC 之含量可依公式(2)計算之。由 EC 對內標準品之 GC 波峰面積比(Y 軸)對其濃度比(X 軸)作圖，其線性迴歸係數  $R^2=0.99$  以上。並計算求得 EC 對內標準品之相對表現因子為 0.42。

### 三、 胺基甲酸乙酯之最低檢出量：

本研究將酒類樣品不經任何前處理，直接取樣注入 GC 中(Method I,) 樣品經濃縮後注入 GC 中(Method II) 及以乙酸乙酯萃出濃縮後注入 GC 中 (Method III) 三種方法分析，採用 FID 為檢測器，訊號設定 FID=1，且 attenuation=1，上述三種分析方法之 EC 最低檢出量分別在 0.5, 0.1 及 0.1  $\mu\text{g/mL}$  左右，如表三所示。

### 四、 添加回收試驗：

表四、表五及表六分別為 5  $\mu\text{g}$  之胺基甲酸乙酯添加至六種酒類飲料(包括蒸餾酒及非蒸餾酒)，然後分別以 Method I、Method II 及 Method III 三種 方法分析之。結果顯示：Method I 之回收率為 95~119%，變異係數(cv%) 均 在 9.2% 以下。Method II 之回收率為 96~112%，變異係數(cv%) 均在 8.9% 以 下。Method III 之回收率為 96~107%，變異係數(cv%) 均在 7.3% 以下。

### 五、 市售酒精性飲料中胺基甲酸乙酯之含量

表七為市售蒸餾酒 21 件及非蒸餾酒 24 件，共 45 件樣品，分別以 Method I 、 Method II 及 Method III 分析定量胺基甲酸乙酯之含量。以 Method I 分析之結果顯示：21 件蒸餾酒中 EC 之含量在 0~2.74  $\mu\text{g/mL}$  和 24 件非蒸餾酒中 EC 含量為 0~1.01  $\mu\text{g/mL}$ ，45 件樣品中 EC 之檢出率達 22%。以 Method II 分析之結果顯示：21 件蒸餾酒中 EC 之含量在 0~2.28  $\mu\text{g/mL}$  和 24 件非蒸餾酒中 EC 含量為 0~1.12  $\mu\text{g/mL}$ ，45 件樣品中 EC 之檢出率達 72%。以 Method III 分析之結果顯示：21 件蒸餾酒中 EC 之含量在 0~2.57  $\mu\text{g/mL}$  和 24 件非蒸餾酒中 EC 含量為 0~1.35  $\mu\text{g/mL}$ ，45 件樣品中 EC 之檢出率達 77%。由上述之結果顯示 Method II 及 Method III 之 EC 檢出率均較 Method I 為高，此乃 Method I 之最低檢出濃度僅為 0.5  $\mu\text{g/mL}$ (如表三所示)，即樣品中 EC 含量若在 0.5  $\mu\text{g/mL}$  以下以便無法檢出以 Method I 便無法檢出。

## 結論

本研究將酒精性飲料樣品，經簡易之前處理，包括：(一)直接取樣 (1 mL) 加入水溶性之內標準品(2-methyl-n-pentanoic acid , IS)後，直接注入GC分析(Method I)；(二)酒類樣品(25 mL)加入內標準品，經濃縮 5~10倍後，直接注入GC分析(Method II)；及(三) 酒類樣品(25 mL) 加入內標準品以乙酸乙酯(v/v=1/2)萃取2~3次，且經濃縮至0.5 mL以下，再注入GC分析 (Method III)。Oven temperature：起始溫100 °C，2分鐘後，以每分鐘3 °C之速率升溫至135 °C後，以每分鐘5 °C之速率升溫至155 °C，再以每分鐘50 °C之速率升溫至300 °C，可以簡單、快速又精確的定量酒精性飲料之胺基甲酸乙酯，分析每一樣品全程僅需20~30分鐘左右。以本研究發展出來的三種方法，檢驗市售各種蒸餾酒及非蒸餾酒共45件之胺基甲酸乙酯含量。結果顯示，中胺基甲酸乙酯之檢出率分別達22、72及77%。以Method I 、Method II 及Method III對上述45件酒類樣品分析，結果顯示45件酒精性飲料中胺基甲酸乙酯之含量分別為0~2.74、0~2.28及0~2.57  $\mu\text{g/mL}$ 。

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**Table 1. Gas chromatographic retention time of some standard candidates**

Compound	Retention time (min) <sup>a</sup>
Guaiacol	11.41
1,3-Butanediol	14.32
2-Methyl-n-pentanoic acid	14.68
3-Methyl-1,5-pentanediol	17.24
1,5-Pantanediol	18.57
1,6-Hexanediol	19.08

<sup>a</sup> CP-Wax column (0.53mmx30m, DF=1.5μm) was used.

Oven condition = 100°C (1min)@ 3°C/min → 135°C @5°C/min → 155°C/min @ 50°C/min → 300°C.

**Table 2. Relative response factor (RRF) and GC retention time (RT) of 2-methyl-n-pentanoic acid to ethyl carbamate**

Compound	RRF <sup>a</sup>	RT <sup>b</sup>
Ethyl carbamate (EC)	0.42	12.73
2-Methyl-n-pentanoic acid (MePA) <sup>c</sup>	1.00	14.68

<sup>a</sup> RRF of 2-methyl-n-pentanoic acid to ethyl carbamate.

<sup>b</sup> CP-Wax column (0.53mmx30m, DF=1.5μm) was used.

<sup>c</sup> Internal standard. Oven condition = 100°C (1min)@ 3°C/min → 135°C @5°C/min → 155°C/min @ 50°C/min → 300°C.

**Table 3. Lowest quantitatively determinable concentration of ethyl carbamate by gas chromatography with FID detector**

Sample preparation*	Concentration ( $\mu\text{g/mL}$ )	Detectability <sup>a</sup>	Recovery (%) <sup>b</sup>	RSD (%) <sup>c</sup>
I	2.5	Yes	102.4	5.8
	1.0	Yes	104.9	6.6
	0.5	Yes	108.1	9.5
	0.25	No	--	--
	0.1	No	--	--
II	2.5	Yes	99.1	4.9
	1.0	Yes	102.8	5.1
	0.5	yes	96.7	7.3
	0.25	yes	107.5	8.9
	0.1	Yes	118.4	12.7
	0.05	No	--	--
III	2.5	yes	97.7	5.7
	1.0	yes	102.3	6.3
	0.5	yes	94.8	7.4
	0.25	Yes	105.6	8.4
	0.1	yes	117.3	13.9
	0.05	no	--	--

\* I = by direct injection method; II = 25mL sample was concentrated to 2-5mL, before injection; III = 25mL sample was extracted with ethyl acetate, and concentrated to 0.5-1 mL, before injection.

<sup>a</sup> FID range = 1, Attenuation = 1.

<sup>b</sup> Average of three analyses.

<sup>c</sup> Coefficient of variation (cv%) .

**Table 4. Recoveries of the spiked ethyl carbamate from various alcoholic liquor or beverages by spitless direct injection GC method (method I )**

Sample	Blank <sup>a</sup> ( $\mu$ g) (A)	Amount added ( $\mu$ g)(B)	Amount found ( $\mu$ g) <sup>b</sup> (C)	Recovery (%) <sup>c</sup>	CV (%) <sup>d</sup>
Whisky 1	ND	5.12	5.79	113.09	7.62
Kaoliang 1	ND	5.12	5.61	109.60	4.38
Medecine wine	ND	5.12	6.14	119.92	9.17
Red wine 1	ND	5.12	5.07	99.02	7.39
Grape wine 1*	3.37	5.12	9.29	109.42	8.27
Plum wine	ND	5.12	4.87	95.12	7.69

<sup>a</sup> Ethyl carbamate in 1 mL whisky.

<sup>b</sup> Average of triplicate analyses.

<sup>c</sup> Recovery (%) =  $(C - A) / B \times 100\%$ .

<sup>d</sup> Coefficient of variation (cv%).

**Table 5. Recoveries of the spiked ethyl carbamate from various alcoholic liquor or beverages by concentration-direct injection GC method (method II)**

Sample	Blank <sup>a</sup> ( $\mu$ g) (A)	Amount added ( $\mu$ g)(B)	Amount found ( $\mu$ g) <sup>b</sup> (C)	Recovery (%) <sup>c</sup>	CV (%) <sup>d</sup>
Whisky 1	1.25	5.12	6.23	97.80	3.47
Kaoliang 1	1.71	5.12	7.17	104.97	5.81
Medecine wine	ND	5.12	4.91	95.90	7.38
Red wine 1	0.38	5.12	6.18	112.36	8.92
Grape wine 1*	4.49	5.12	9.33	97.09	5.67
Plum wine	1.39	5.12	6.71	103.07	2.19

<sup>a</sup> Ethyl carbamate in 25 mL whisky.

<sup>b</sup> Average of triplicate analyses.

<sup>c</sup> Recovery (%) =  $(C - A) / B \times 100\%$ .

<sup>d</sup> Coefficient of variation (cv%).

**Table 6. Recoveries of the spiked ethyl carbamate from various alcoholic liquor or beverages by solvent extraction-spitless direct injection GC method (method III)**

Sample	Blank <sup>a</sup> ( $\mu$ g) (A)	Amount added ( $\mu$ g)(B)	Amount found ( $\mu$ g) <sup>b</sup> (C)	Recovery (%) <sup>c</sup>	CV (%) <sup>d</sup>
Whisky 1	1.32	5.12	6.86	106.52	3.89
Kaoliang 1	1.36	5.12	6.25	96.45	5.64
Medecine wine	ND	5.12	5.48	107.03	7.28
Red wine 1	ND	5.12	4.98	97.27	6.89
Grape wine 1*	4.58	5.12	15.02	103.29	7.34
Plum wine	1.12	5.12	6.37	102.08	5.43

<sup>a</sup> Ethyl carbamate in 25 mL whisky.

<sup>b</sup> Average of three analyses.

<sup>c</sup> Recovery (%) =  $(C - A) / B \times 100\%$ .

<sup>d</sup> Coefficient of variation (cv%).

**Table 7. Ethyl carbamate levels found in some alcoholic beverage**

Sample	EC content (μg/mL)		
	Method I	Method II	Method III
<u>Distilled spirit</u>			
Whisky 1	1.02	1.25	1.32
Whisky 2	ND	0.78	0.85
White liquor 1	ND	ND	ND
Kao liang 1	1.34	1.71	1.25
Kao liang 2	0.93	0.83	0.79
Kirin whisky	ND	0.26	0.31
XO-1	ND	0.20	0.18
XO-2(J)	ND	ND	0.14
VSOP-A	ND	0.32	0.41
VSOP-B	ND	0.24	0.31
Drambule-B	2.74	2.28	2.91
Kliporiff-c	ND	0.56	0.62
Drambule-d	2.43	1.82	2.57
Martell-e	ND	ND	ND
Pushkin-f	ND	ND	ND
Smirnoff-g	ND	ND	ND
Cocorico-h	1.46	1.72	1.28
J&B-I	2.01	1.23	1.59
Rice wine 1	ND	0.21	0.15
Rine wine 2	ND	0.20	0.13
Rine wine heads	ND	0.18	0.14
<u>Non-distilled spirit</u>			
Plum wine 1	ND	0.59	0.72
Plum wine 2	ND	0.43	0.35
Plum wine 3	ND	0.23	0.37
Red wine 1	ND	ND	0.12
Red wine 2	ND	ND	ND
Red wine 3	ND	ND	0.13
Red rose 1	ND	ND	ND
Red rose 2	ND	ND	0.13
Red rose 3	ND	ND	0.11
White wine 1	ND	ND	0.14
White wine 2	ND	ND	0.11
White wine 3	ND	ND	0.17
Yellow wine 1	ND	0.62	0.84
Yellow wine 2	ND	0.54	0.73
Shao-Hsing wine 1	0.88	1.12	1.35
Shao-Hsing wine 2	0.74	0.69	0.85
Taiwan beer 1	ND	0.28	0.34
Taiwan beer 2	ND	0.37	0.41
Beer 3	ND	0.22	0.37
Beer 4	ND	0.38	0.42
Medecine wine 1	ND	ND	ND
Medecine wine 2	1.01	0.82	0.75
Medecine wine 3	ND	0.15	0.19
Medecine wine 4	ND	ND	0.11
Detectability (%)	22.22%	71.79%	76.92%

<sup>a</sup> Average of two analyses.